

## Introduction

I was inspired by the *Taboo* game, supplied in the QCA/KS3 Strategy Booster lessons for Year 9, to produce a similar game for use with Year 11 pupils to help them to revise some difficult key words. I printed the *Taboo* cards off then replaced them into the feeder tray to print the *Taboo* labels on the back, making them look more like a set of game cards. Year 11 pupils enjoyed the game and I hope they will be of some use to other teachers.

(Andrea Dart, Westhoughton High School, Wigan)

## Running the activity

*Taboo* is a card game played in two teams. Individuals take it in turns to describe words to their team within a time limit. The only catch is that for each word that the individual is given to describe there are a list of words they are not allowed to include within their description. For example, if the person is trying to describe the word 'distillation' to their team, words they are not allowed to use are 'purify', 'separate', 'boiling point' and 'condense'. The team at the end of the time you have allotted for the game that has managed to explain the most words wins.

## Safety

Not applicable.

## More ideas

The Booster lessons Angela refers to in her introduction can be found on the DfES Standards website, <http://www.standards.dfes.gov.uk/keystage3/publications/> There are a number of lesson plans and resources available in Word format, which you can amend if you wish before printing out.

## Learning outcomes

- Recap of chemistry topics content

## Where the activity fits in

Chemistry (ages 14-16)

## Skills

Recall, vocabulary.

## Acknowledgements

Many thanks to Andrea Dart for sharing this resource.

COVALENT

---

share  
electrons  
non-metal

NEUTRALIS-  
ATION

---

acid  
alkali  
neutral

OXIDATION

---

oxygen  
react  
combustion  
electrons

REDUCTION

---

oxygen  
reduce  
electrons

ELECTROLYSIS

---

electricity  
separate  
ions  
aluminium

DISTILLATION

---

purify  
separate  
boiling point  
condense

## PERIOD

---

atoms  
periodic table  
electron shells  
across

## IONIC BOND

---

metal  
non-metal  
strong  
ions

## GROUP

---

periodic table  
elements  
properties  
down

## NON-METAL

---

poor conductor  
brittle  
gas

## ELECTRON

---

atoms  
negative  
electron shells  
reactivity

## METAL

---

conductor  
shiny  
ductile  
malleable

ACID

---

pH  
alkali  
neutral  
indicator

ALKALI

---

pH  
acid  
neutral  
indicator

pH

---

acid  
alkali  
neutral  
1-14

FRACTIONAL  
DISTILLATION

---

purify  
oil  
boiling point

HABER  
PROCESS

---

ammonia  
nitrogen  
hydrogen

SALT

---

compound  
metal  
non-metal

HYDROCARBON

---

hydrogen  
carbon  
atoms  
oil

ALKANE

---

hydrocarbon  
carbon  
saturated  
single bonds

ALKENES

---

hydrocarbon  
carbon  
unsaturated  
double bond

POLYMER-  
ISATION

---

polymer  
monomer  
plastic

COMBUSTION

---

burn  
oxygen

THERMAL  
DECOMPOSITION

---

heat  
break down

## ELEMENT

---

pure  
atoms  
periodic table

## MIXTURE

---

atoms,  
easy to  
separate,  
not bonded

## COMPOUND

---

atoms bonded  
molecule  
hard to  
separate

## ATOM

---

element  
protons,  
neutron &  
electrons

## PROTON

---

atom  
period number  
positive  
nucleus

## NEUTRON

---

atoms  
atomic mass  
no charge  
nucleus

